



C-Glycosides and C-Disaccharide Precursors Through Carbonylative Stille Coupling Reactions

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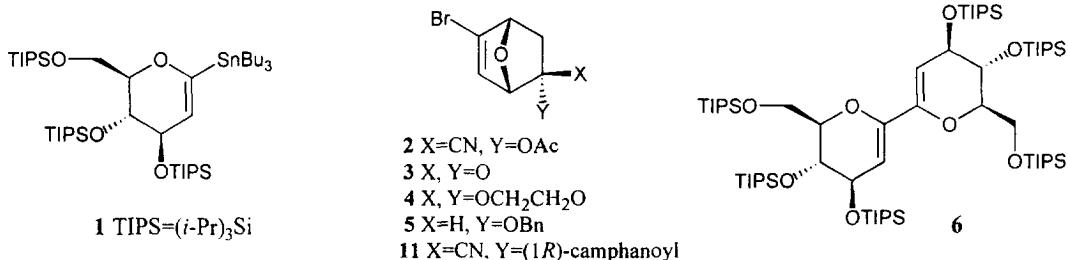
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Abstract: Under CO atmosphere and in the presence of $Pd_2(dba)_3$ and Ph_3As a suitably protected 1-stannylglucal derivative could be carbonylated and coupled to 5-bromo-7-oxabicyclo[2.2.1]hept-5-en-2-yl derivatives. The carbonylative Stille coupling was also successful between 1-iodoglucals and tributyl(vinyl)stannane or tributyl(fur-2-yl)stannane. A cross-conjugated dienone was also obtained through coupling of a 1-stannylglucal with a 1-iodoglucal derivative.

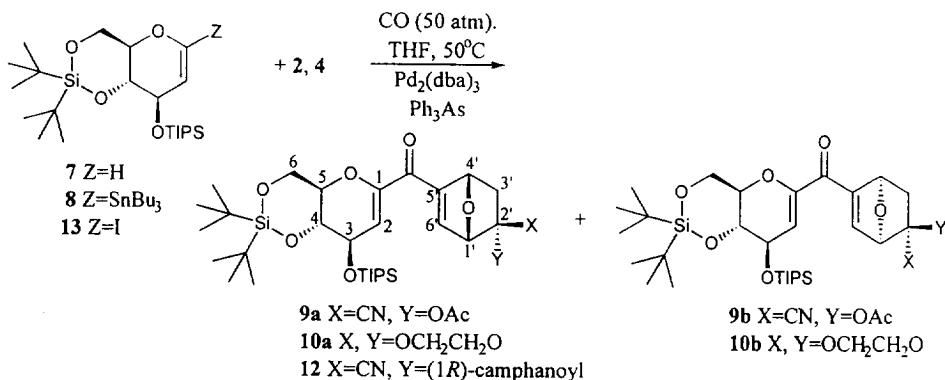
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Oligosaccharides play vital roles in intercellular communication and all cell mediated processes.^{2,3} Mimics of oligosaccharides such as the C-disaccharides⁴ may or may not imitate the physical and biological properties of the corresponding O-glycosides⁵; they present the advantage to resist acidic and enzymatic hydrolysis. This constitutes for these C-glycosides a potential as glycosidase inhibitors and as non-hydrolysable epitopes. Since the first synthesis of a β -(1→6)-C-disaccharide by Rouzaud and Sinay⁶ several approaches to the C-disaccharides have been proposed.^{4,7} They usually imply multiple step transformations of two sugars, one of them being implemented by a carbon function, subsequent coupling through C-C bond formation and further transformations. We disclose here our preliminary results on a new approach that generates the C-link between two hexose precursors in a single step. The method relies on the carbonylative Stille coupling reaction⁸ between 1-stannylglycals and 5-bromo-7-oxabicyclo[2.2.1]hept-5-en-2-yl derivatives obtained readily from the Diels-Alder adducts of furan to 1-cyano-vinyl esters.^{9,10} Earlier attempts to carry out anomeric alkylcarbonylations of glycosyl bromides involved the reactions of the corresponding (pentacarbonyl)glycosylmanganese with Michael acceptors under very high pressure.^{11,12}

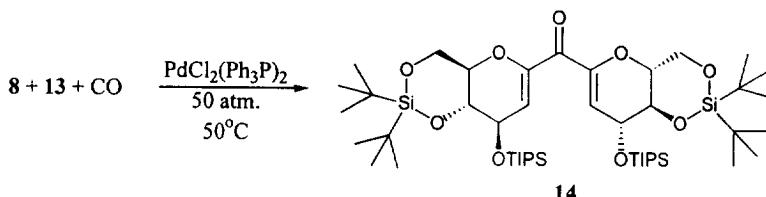


In a first series of experiments we treated the known stannylated glucal derivative **1**¹³ with racemic 5-bromo-7-oxabicyclo[2.2.1]hept-5-en-2-yl derivatives (\pm)-**2** - (\pm)-**5** under CO atmosphere (1-150 atm.) in the presence of a variety of palladium complexes, ligands and solvents^{8,14} but we failed to detect any product of carbonylative Stille coupling. In some cases, and as expected for reactions in the absence of CO,¹⁵ we

observed the formation of **6** resulting from the Stille homocoupling of **1**. We suspected that the three bulky TIPS protective groups in **1** could impede the carbonylation for steric or/and conformational reasons. We thus converted tri-O-acetyl-D-glucal into **7** via methanolysis (MeONa/MeOH , 25°C , 2 h) of the acetates, followed by silylation first with $(t\text{-Bu})_2\text{Si}(\text{OSO}_2\text{CF}_3)_2/2,6\text{-lutidine}$ in anhydrous DMF (-50°C , 10 h), then with $(i\text{-Pr})_3\text{SiCl}/\text{imidazole}$ (DMF, 50°C , 24 h). Treatment of **7** with $t\text{-BuLi}$,¹⁶ then with Bu_3SnCl furnished **8** (80% based on tri-O-acetyl-D-glucal).

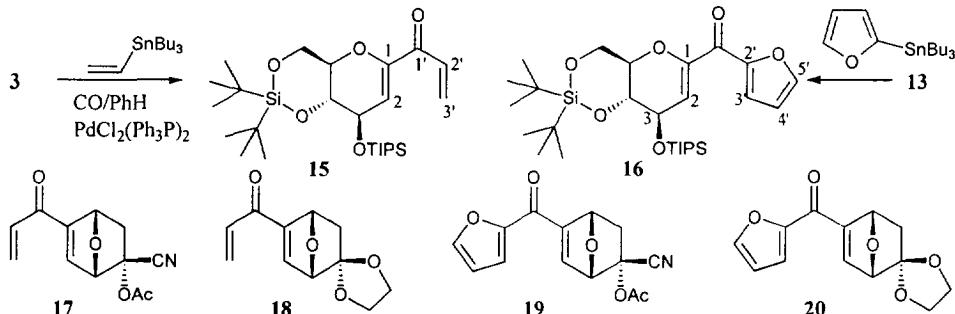


In the presence of 5-10 mol% of $\text{Pd}_2(\text{dba})_3$ (dibenzylideneacetone-palladium), 10-20 mol% of Ph_3As in anhydrous THF and under 50 atmospheres of CO at 50°C , 1:1 mixtures of **8** + $(\pm)\text{-2}$ and **8** + $(\pm)\text{-4}$ led to 1:1 mixtures of **9a/9b** and **10a/10b**, respectively, in 30-55% yield. Applying this process to the enantiomerically pure derivative **11**, dienone **12** was isolated (47%).¹⁷ Replacement of Ph_3As for Ph_3P failed to produce products of carbonylative Stille coupling **9** and **10**. When **8** was treated with 0.5 equiv. of I_2 in CH_2Cl_2 (what generates 0.5 equiv. of 1-iodoglucal **13**) and then pressurized with CO (50 atm.) in the presence of 30% mol of $\text{PdCl}_2(\text{Ph}_3\text{P})_2$, dienone **14**¹⁸ was obtained in 65% yield.¹⁹



In order to test whether other C-glycosides could be generated through the carbonylative Stille heterocoupling reaction we treated the 1-iodo-D-glucal derivative **13** with commercially available (tributyl)vinylstannane and (tributyl)(fur-2-yl)stannane under CO pressure (50 atm) in benzene, at 60°C and in the presence of $\text{PdCl}_2(\text{Ph}_3\text{P})_2$. The expected cross-conjugated dienes **15**²⁰ and **16**²¹ were isolated in 71 and 91% yield, respectively. Under similar conditions (THF, 50°C , $\text{Pd}(\text{Ph}_3\text{P})_4$) 5-bromo-7-oxanorbornenes $(\pm)\text{-2}$ and $(\pm)\text{-4}$ were coupled with CO and the same stannanes giving the corresponding products of carbonylative heterocoupling **17**, **18**, **19** and **20** in 71, 70, 90 and 90% yield, respectively.

Work is underway in our laboratory to convert the new C-glycosides described in this report to a variety of disaccharide mimics and other compounds of biological interest.²²



Acknowledgments. We thank the Swiss National Science Foundation and "Les Laboratoires Fournier" Dijon, France, for generous financial support.

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17. Data for **12**: $[\alpha]_{589}^{25} = -66$, $[\alpha]_{577}^{25} = -69$, $[\alpha]_{546}^{25} = -79$, $[\alpha]_{435}^{25} = -123$, $[\alpha]_{405}^{25} = -115$ ($c=1.44$, CHCl_3); $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 6.87 (d, $^3J = 1.8$, HC(6')), 5.85 (d, $^3J=2.5$, HC(2)), 5.47 (d, $^3J=1.8$, HC(1')), 5.38 (dd, $^3J=4.9$, 1.0, HC(4')), 4.56 (dd, $^3J=7.0$, 2.5, HC(3)), 4.26 (dd, $^2J=10.2$, $^3J=4.9$, HC(6)), 4.12-4.00 (m, $^2J=10.2$, 10.2, 7.0, 4.7, HC(4), HC(6)), 3.94 (ddd, $^3J=10.2$, 4.9, 4.7, HC(5)), 2.91 (dd, $^2J=13.1$, $^3J=4.9$, $\text{H}_{\text{exo}}\text{C}(3')$), 1.88 (dd, $^2J=13.1$, $^3J=1.0$, $\text{H}_{\text{endo}}\text{C}(3')$), 2.34, 2.04-1.91, 0.96 (4H camphanate), 1.09, 1.04, 0.96 (3 Me camphanate), 1.38-1.10 (m, OTIPS), 1.08, 1.00 (2s, 2 Si(*t*-Bu)).
18. Data for **14**: IR (film) v: 2935, 2890, 2865, 2360, 2340, 1470, 1160, 1110, 1070 cm^{-1} ; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 5.82 (d, $^3J = 2.4$, HC(2)), 4.54 (dd, $^3J=7.1$, 2.4, HC(3)), 4.27 (ddd, $^3J=10.3$, 5.0, 4.9, HC(5)), 4.08-3.99 (m, $^2J=10.3$, $^3J=10.3$, 7.1, 5.0, HC(6), HC(4)), 3.89 (dd, $^2J=10.3$, $^3J=4.9$, HC(6)), 1.12-0.98 (m, OTIPS, 2 Si(*t*-Bu)); $^{13}\text{C-NMR}$ (100.61 MHz, CDCl_3) δ_{C} : 182.4 (s, CO), 148.4 (s, C(1)), 115.5 (d, $^1J(\text{C},\text{H})=169$, C(2)), 76.4, 73.2, 71.0 (3d, $^1J(\text{C},\text{H})=150$, C(3), C(4), C(5)), 65.7 (t, $^1J(\text{C},\text{H})=147$, C(6)), 27.3, 26.8 (3q, $^1J(\text{C},\text{H})=125$, $\text{Si}(\text{C}(\text{CH}_3)_3)_2$, $\text{SiCH}(\text{CH}_3)_2$), 22.7, 19.8 (2s, $\text{Si}(\text{C}(\text{CH}_3)_3)_2$), 12.3 (d, $^1J(\text{C},\text{H})=118$, $\text{SiCH}(\text{CH}_3)_2$).
19. Product of carbonylative homocoupling of glycal derivatives have been obtained by treatment of 1-lithiated tri-O-benzyl-D-glucal with dimethyl carbonate.¹⁶
20. Data for **15**: $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 6.86 (dd, $^3J=17.2$, 10.5, HC(2')), 6.40 (dd, $^2J=1.8$, $^3J=17.2$, $\text{H}_{\text{trans}}\text{C}(3')$), 5.84 (d, $^3J=2.5$, HC(2)), 5.79 (dd, $^2J=1.8$, $^3J=10.5$, $\text{H}_{\text{cis}}\text{C}(3')$), 4.56 (dd, $^3J=7.0$, 2.5, HC(3)), 4.31 (dd, $^2J=10.3$, $^3J=5.0$, HC(6)), 4.10-4.03 (m, $^2J=10.3$, $^3J=10.3$, 7.0, 4.9, HC(6), HC(4)), 3.92 (ddd, $^3J=10.3$, 5.0, 4.9, HC(5)), 1.13-0.91 (m, OTIPS, 2 Si(*t*-Bu)).
21. Data for **16**: $[\alpha]_{589}^{25} = -40$, $[\alpha]_{577}^{25} = -41$, $[\alpha]_{546}^{25} = -48$, $[\alpha]_{435}^{25} = -92$, $[\alpha]_{405}^{25} = -121$ ($c=1.48$, CHCl_3); IR (film) v: 2940, 2890, 2865, 2360, 2340, 1465, 1060 cm^{-1} ; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} : 7.64 (dd, $^3J=1.7$, $^4J=0.7$, HC(5')), 7.33 (dd, $^3J=3.6$, $^4J=0.7$, HC(3')), 6.52 (dd, $^3J=3.6$, 1.7, HC(4')), 5.99 (d, $^3J=2.5$, HC(2)), 4.57 (dd, $^3J=7.1$, 2.5, HC(3)), 4.33 (dd, $^2J=10.2$, $^3J=4.9$, HC(6)), 4.15-4.07 (m, $^2J=10.2$, $^3J=10.3$, 7.1, 5.0, HC(4), HC(6)), 3.99 (ddd, $^3J=10.3$, 5.0, 4.9, HC(5)), 1.25-0.90 (m, 2 (*t*-BuSi), OTIPS); $^{13}\text{C-NMR}$ (100.61 MHz, CDCl_3) δ_{C} : 174.3 (s, CO), 150.6 (s, C(1)), 148.8 (s, C(2')), 147.4 (d, $^1J(\text{C},\text{H})=200$, C(5')), 121.4 (d, $^1J(\text{C},\text{H})=180$, C(2)), 113.6 (d, $^1J(\text{C},\text{H})=164$, C(3')), 112.1 (d, $^1J(\text{C},\text{H})=178$, C(4')), 76.6, 73.6, 71.1 (3d, $^1J(\text{C},\text{H})=150$, C(3), C(4), C(5)), 65.7 (t, $^1J(\text{C},\text{H})=147$, C(6)), 27.3, 26.8, 18.0 (3q, $^1J(\text{C},\text{H})=125$, $\text{Si}(\text{C}(\text{CH}_3)_3)_2$, $\text{SiCH}(\text{CH}_3)_2$), 22.7, 20.9 (2s, $\text{Si}(\text{C}(\text{CH}_3)_3)_2$), 13.8 (d, $^1J(\text{C},\text{H})=117$, $\text{SiCH}(\text{CH}_3)_2$).
22. Part of this work has been disclosed at the 2nd ECCM, Lago di Garda, Italy. July 16-19, 1996.

(Received in France 14 October 1996; accepted 28 November 1996)